Electrostatic self-assembly of concentric foam shells for cryogenic laser targets

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Typical parameters for droplet centering



$$E \approx 3 \cdot 10^4 V_{rms}/m$$

DMA & silicone oil are density matched to within ~0.2%.

Experimental apparatus with holder



Original "aquarium"

Centering phenomenology

Outer shell: 55 µl DMA droplet w/Span 80

 $K = 37.8 \& O_2 \approx 10^{-3} \text{ S/m}$

Inner droplet: 45 µl silicone oil/TECE $\kappa_1 = \kappa_3 = 2.5$ & $\sigma_1 \approx \sigma_3 \approx 2 \cdot 10^{-6}$ S/m

$E \sim 2 \cdot 10^4 \,\mathrm{V_{rms}/m}$ @ $f \sim 20 \,\mathrm{kHz}$ $\rightarrow 200 \,\mathrm{kHz}$



E-field centering depends on frequency. 4

E-field induced droplet distortion



Droplets elongate in the electric field but centering time is reduced -- might we exploit this effect to accelerate centering?

For this experiment, centering time reduced from 80 s to ~45 s.

Why does electrostatic self-assembly work?



Energy argument

When inner droplet is centered, it is in equilibrium.

This equilibrium will be either a MIN or a MAX of the electrostatic energy.

Therefore, equilibrium will be either **stable** or **unstable**.

Image dipole interpretation of centering:



 $\kappa_2 > \kappa_1$ guarantees stability; κ_3 has no influence!

7



9 _m/pixel; 1.5 fps

Image analysis based on ellipsoid fit: ellipsoidal distortion in terms of a/b



9 m/nivel: 1.5 fm Glass holder plate causes both vertical mis-alignment (~45 μ m) & oblate vertical distortion of outer shell (a/b ~ 1.01).

A better idea: density gradient suspension: the silicone oil parfait



Results with gradient suspension



$t \approx 0 \text{ s}$

$t \approx 60 \text{ s}$

DMA + Mordant blue dye: 0.1 g/liter elect. conductivity: $\sigma \approx 2.5 \cdot 10^{-3}$ S/m $E \approx 3.6 \cdot 10^4$ V_{rms}/m @ 3.8 MHz

DMA + Mordant blue dye shell



∆centroid: <20 μm elongation: <1%





How good is this shell?

"unwrapped" power spectral density

oil/DMA/oil droplet: diameter = 4.76 mm wall = 260 µm

outer surface: ~4.7 μm rms

inner surface: ~4.5 μm rms



Dominant mode: m = 2

Requirements for foam shell formation



Outer shell will contain the monomer in solution (e.g., resorcinol/formaldahyde), which is likely to be electrically conductive

Inner droplet: inert liquid mandrel

Shielding due to higher shell conductivity increases frequency requirement of *E*: $\sim 1 \text{ kV} @ \geq 1 \text{ MHz}$ (fortunately at very low current)

Electrical properties of expt'l liquids

constituent	conductivity ¶"	dielectric const. κ
Suspension liquid: Silicone oil	<10 ⁻⁶ S/m	~2.5
Outer shell:		
DMA (N,N-dimethylacetamide)	~2•10 ⁻⁴ S/m	37.8
DMA+Mordant blue dye (0.1 g/l)	~2.5•10 ⁻³ S/m	37.8
Resorcinol/formaldehyde	~2.53•10 ⁻² S/m	???
Inner droplet: Silicone oil	<10 ⁻⁶ S/m	~2.5

Estimates for critical frequency for IFE shell:

$$f_{\text{shielding}} = \frac{1}{2\pi} \frac{2\sigma_2 d}{\varepsilon_3 R} =$$

{ ~200 kHz (DMA) ~3.8 MHz (DMA+dye) ~40 MHz for R/F ¹⁵

Challenge for the chemists:

Identify polymer foam chemistries that minimize or avoid ionic salts.

- Use radical initiators?
- Avoid surfactants if possible.
- Photo-initiated polymerization would be ideal.
- Keep monomer solution conductivity $\leq 10^{-3}$ S/m: ion salt molarity of ~1 mM yields $\sigma \sim 10^{-3}$ S/m.

Important questions

- How close must densities be matched? <0.2%?
- How rapidly is concentricity approached? ~60 seconds
- Is particle elongation a problem? a/b < 1.01
- Does use of density gradient suspension avoid need for surfactant? Apparently yes!
- What geometry to use for centering droplets?
- Will Joule heating-induced convection obstruct centering in early phase of polymerization?
- Are there polymer foam-forming solutions with electrical conductivity in the range of ~10⁻³ S/m?

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